A TRITERPENOID SAPONIN FROM THLADIANTHA HOOKERI VAR. PENTADACTYLA

Rui-lin Nie, Takahiro Tanaka,* Masazumi Miyakoshi,* Ryoji Kasai,* Toshinobu Morita,* Jun Zhou and Osamu Tanaka**†

Kunming Institute of Botany, Chinese Academy of Science, Kunming Yunnan, China; *Institute of Pharmaceutical Sciences, Hiroshima University School of Medicine, 1-2-3, Kasumi, Minami-ku, Hiroshima 734, Japan

(Received 20 September 1988)

Key Word Index—Thladiantha hookeri var. pentadactyla; Cucurbitaceae; Wu-ye-chi-pa; saponin; Chinese folk medicine; gypsogenin.

Abstract—A new triterpenoid saponin was isolated from the tubers of *Thladiantha hookeri* C. B. Clarke var. pentadactyla in a high yield (9.7%). On the basis of chemical and spectral evidence, the structure was determined as β -xylopyranosyl-(1 \rightarrow 3)- β -xylopyranosyl-(1 \rightarrow 4)- α -rhamnopyranosyl-(1 \rightarrow 2)- β -xylopyranosyl ester of 3-O- β -galactopyranosyl-(1 \rightarrow 2)- β -glucuronopyranosyl-gypsogenin. The solubilizing effect of this saponin is also described.

INTRODUCTION

In continuation of studies on Chinese cucurbitaceous folk medicine, the present paper reports the isolation and structural determination of a new saponin named thladioside-H1 (1) from tubers of *Thladiantha hookeri* C. B. Clarke var. *pentadactyla* Cogn. (Chinese name: Wuye-chi-pa).

RESULTS AND DISCUSSION

An ethanolic extract of the tubers was separated as described in the Experimental, affording a new saponin named thladioside-H1 (1) in a surprisingly high yield (9.7%).

Compound 1 was hydrolysed with acid to yield glucuronic acid(GlcUA), galactose(Gal), xylose(Xyl) and rhamnose(Rha) as sugar components. On enzymatic hydrolysis with crude hesperidinase [1], compound 1 gave gypsogenin (2) [2, 3] as an aglycone. The ¹H and ¹³C NMR spectra (Tables 1 and 2) of compound 1 showed the presence of six monosaccharide units. On comparison of the ¹³C NMR spectrum of compound 1 with that of compound 2, the glycosylation shifts [4] were observed for C-2, C-3, C-23 and C-28, indicating that 1 was a bisdesmoside of 2 with glycosyl linkages at both the 3-hydroxyl and 28-carboxyl groups.

Since the aldehyde group would perturb the structure elucidation, compound 1 was converted into a relatively stable saponin (3) of hederagenin (4) [2, 5] by reduction of the aldehyde group to a primary alcohol with sodium borohydride in water.

On selective cleavage of the ester glycoside linkage at C-28 with LiI-2,6-lutidine in methanol [6], compound 3 gave a monodesomoside (5) and a methyl tetrasaccharide (6), the former of which was purified as a monomethyl ester (7) by mild treatment with methanolic hydrogen chloride. Acid hydrolysis of compound 7 yielded GlcUA

and Gal. Anomeric configurations of both the glycosyl linkages were assigned as β by the coupling constant of the anomeric proton signals. The location of the methyl

Table 1. 13 C NMR chemical shifts of aglycone moseties in C_4D_4N

- -									
C	1	2	7	8	9	10	11		
1	38.0	38.4	38.5	38.8	38.6	38.7	38.6		
2	25.1	27.0	25.9	258	27.0	27.1	27.0		
3	82.0	71.6	82.3	82.3	71.5	71.7	71 6		
4	54.9	56.2	42.0°	42.3	56.2	56.3	56.3		
5	48.6ª	48.0ª	48.0ª	48.2*	48.0°	48.0	48.0		
6	20.6	21.0	18.1	18.2	21.1	21.3	21.2		
7	32.5	32.5	32.8	32.9	32.4 ^b	32.5	32.5ª		
8	40.1	40.0	39 7	39.8	40.2	40.2	40.2		
9	47.8ª	47.7°	47.6°	47.8	47.8*	48.0	48.0		
10	36.2	36.2	36.8	37.0	36.1	36.2	36.2		
11	23.6	238	23.8	23.8	23.6	23.7	23.6		
12	122.5	122.2	122.5	122.6	122.4	122.6	122.5		
13	144.0	144.8	144.5	144.8	144.2	144.2	144 1		
14	42 2	42.2	42 1	42.3	42.4	42.4	42.4		
15	28.4	28.2	28.3	28 5	28.7	28.5	28.5		
16	23.6	23.8	23.8	23.9	23.2	23.4	23.2		
17	47.2	46.6 ^b	46.6 ^b	46.8	47.2	47.3	47.2		
18	41.9	419	42.1°	423	41.9	42.0	41.9		
19	46.7	46.5 ^b	46 4 ^b	46.8	46.4	46.5	46.3		
20	30.7	30.9	30.9	31.0	30.7	30.8	30.8		
21	34.1	34.2	34.2	34.3	34.1	34.2	34.0		
22	32.5	33.1	33.2	33.3	32.5 ^b	32.5	32.7ª		
23	210.0	207.1	64.1	64.5	207.1	2073	207.8		
24	10.0	9.6	13.5	137	9.6	9.8	9.7		
25	15.7	15.7	16.0	16.1	15.7	15.9	15.8		
26	17.3	173	17.4	17.6	17.3	17.4	17.4		
27	25.8	26.1	26.2	26.3	25.8	25.9	25.9		
28	176.4	180.0	180 2	180.2	176.4	176.5	176.4		
29	33 1	33.2	33.2	33.3	33.1	33.1	33.1		
30	23.6	23.8	23.8	23.9	23.6	23.7	23.6		

[†]Author to whom correspondence should be addressed.

a, b, c Assignments may be interchanged in each column.

1712 R Nie et al

Table 2. 13C NMR chemical shifts of sugar moieties in C₅D₅N

	1	7		8	9	10	11
3-O-GlcUA-1	103 4	104.2	3-O-Glc-1	103 9			
2	83 8	83 3	2	83 9			
3	77 4	77 Oª	3	78 0			
4	728	72 7	4	710			
5	77 4	76 7ª	5	78 5			
6	172 3	1704	6	62.7			
Gal-1	106.0	106.6	Gal-1	105 1			
2	74 2	74.4	2	74.4			
3	750	75 0	3	750			
4	70.0	69 6	4	710			
5	77 4	77 5	5	77 O			
6	620	61 4	6	61 5			
28-O-Xyl-1	95 1				953	952	95 2
2	78 0				78 3	78.6	78 0
3	758				754	75.9ª	75.9ª
4	70 7				708	709	708
5	67.2				67.2	67 2 ^b	66 9 ^b
Rha-1	101 2				101 4	101 3	101 2
2	72.5				72 1	72 5	72.5
3	71.5				72.5	71 3	71 6
4	849				73 7	85 1	848
5	68 1				69.8	68 4	68 2
6	18 5				18.7	186	18.5
Xyl-1	105 7					107 6	105 7
2	75 O*					76 2ª	75 1ª
3	870					77 7	87 1
4	68 8					70 9	68.9
5	67 2					67 5 ^b	66 7 ^t
Xyl-1	106 8						106.2
2	74 8 ^a						75 1ª
3	77 4						77.5
4	70 7						70.8
5	67.2						67.3

a,b,c Assignments may be interchanged in each column

ester group of compound 7 at the GlcUA unit was confirmed by the ¹³C NMR spectrum (Table 1) Compound 7 was reduced with sodium borohydride or sodium borodeuteride in ethanol [7], whereby the GlcUA unit was converted into a corresponding neutral hexose unit or deuterated neutral hexose unit, affording 8 and 8-2H₂, respectively. The methylation-analysis [8] of compound 8-2H2, afforded 1,5-d1-O-acetyl-2,3,4,6-tetra-O-methylhexitol and 1,2,5-tri-O-acetyl-6-dideutero-3,4,6tri-O-methylhexitol, which were proved to be associated with terminal galactopyranosyl and 2-linked glucopyranosyl-d2 residues, respectively, by GC-MS. The coupling constant of both the anomeric proton signals indicated the β -anomeric configuration. It follows that the structure of 5 was assigned as the 3-O- β -galactopyranosyl- $(1\rightarrow 2)$ - β -glucuronopyranoside of compound 4 and this structure was supported by the ¹³C NMR spectrum

Compound 1 was treated with β -glucuronidase and the resulting neutral prosapogenin mixture was separated by CC to give three compounds (9–11). The FD mass spectra of compounds 9–11 exhibited molecular ion peaks as a cationized cluster ion at m/z 771 $[M+Na]^+$ [M=2-(Xyl-Rha)], 903 $[M+Na]^+$ [M=2-(2Xyl-Rha)], 1035

 $[M+Na]^+$ [M=2-(3Xyl-Rha)], respectively, so that compound 11 was assigned as 3-desglyco-prosapogenin of 1.

The EI-mass spectrum of a peracetate of compound 9 exhibited fragment ions at m/z 273 [(Rha)Ac₃]⁺ and 489 [(Rha-Xyl)Ac₅], which, coupled with the ¹H(anomeric proton) and ¹³C NMR signals due to the sugar moiety, led to the formulation of 9 as the α -rhamnopyranosyl- $(1\rightarrow 2)$ - β -xylopyranosyl ester of compound 2

The EI-mass spectrum of a peracetate of compound 10 showed fragment ions at m/z 259[(XyI)Ac₃]. 489[(XyI-Rha)Ac₅] and 705[(XyI-Rha-XyI)Ac₇] Compound 10 was treated with LiI-2,6-futidine in methanol, yielding a methyl trisaccharide (12). The methylation analysis of compound 12 revealed the presence of terminal xylopyranosyl and 2-linked xylopyranosyl and 4-linked rhamnopyranosyl residues. This evidence coupled with the formulation of compound 9 led to the structure of compound 10 as shown in Chart 1 The anomeric configuration of each glycosyl linkage of 10 was substantiated by the ¹H and ¹³C NMR spectra

The methylation analysis of compound 6 obtained from 3 (vide supra) proved the presence of terminal, 2-

linked, 3-linked xylopyranosyl and 4-linked rhamnopyranosyl residues. Based on this evidence and the formulation of 9 and 10 together with the elucidation of the anomeric configuration by the ¹H and ¹³C NMR spectra, the structure was assigned to compound 11.

Accordingly, the structure of compound 1 was established as $28-O-\beta$ -xylopyranosyl- $(1\rightarrow 3)-\beta$ -xylopyranosyl- $(1\rightarrow 4)-\alpha$ -rhamnopyranosyl- $(1\rightarrow 2)-\beta$ -xylopyranosyl ester of $3-O-\beta$ -galactopyranosyl- $(1\rightarrow 2)-\beta$ -glucuronopyranosyl-gypsogenin.

In our serial studies on surfactive properties of natural oligo-glycosides, we have disclosed that glucuronide saponins such as ginsenoside-Ro (oleanolic acid saponin of Ginseng root) [9] and glycyrrhizin [10] strongly increase the water solubility of saikosaponins, active principles of Bupleuri radix. The essential role of the glucuronide moiety in this solubilizing effect was also revealed. In the present study, this specific solubilizing effect was observed with 1, the saponin with a glucuronide unit (Table 3). Compound 1 also solubilized Yellow OB (a water insoluble synthetic dye) and increased the water solubility of saponin A from pericarps of Sapindus mukurossi which is sparingly soluble in water (Tables 4, 5).

EXPERIMENTAL

NMR spectra were measured in C_5D_5N -TMS, ¹H NMR at 100 or 270 MHz and ¹³C NMR at 25 and 67.8 MHz. The solvents used for spectral determination were; MeOH([α]_D); Nujol(IR), unless otherwise stated. Acid hydrolysis of saponins followed by identification of the resulting monosaccharides and the methylation analysis of sugar moiety by GC-MS were carried out as described in the previous paper [8]. For CC, silica gel 60 was used. The solvent systems for CC was all homogeneous.

Plant material. Thladiantha hookeri C. B. Clarke var. pentadactyla Cogn. was collected at Lufeng (Yunnan), China in June 1970 and identified by Emeritus Professor C.-Y. Wu (Kunming In-

stitute of Botany, Chinese Academy of Science). A voucher specimen has been deposited in the Herbarium of this Institute.

Extraction and separation The dried and powdered tubers (2 kg) were extracted with hot EtOH. After removal of the solvent by evapn, the residue was separated into H₂O-soluble and insoluble portions by treatment with H₂O and an aq. soln of the water-soluble portion was extracted with Et₂O and then with n-BuOH satd with H₂O. After removal of solvents by evapn, the n-BuOH extract (210 g) was proved to consist of an almost homogeneous saponin, compound 1 by TLC on silica gel (CHCl₃-MeOH-H₂O, 13.7 2, lower layer), visualized by spraying with 10% ethanolic H₂SO₄ followed by heating at 110° and on a reverse phase plate, HPTLC RP-3 Merck [solvent 70% MeOH, visualized in the same way as above] This fraction

Table 3. Solubilizing effect on saikosaponin-a in water at 37°

Compound (1 mg/ml)	Solubility of saikosaponin-a (mg/ml)
None	0.1
Ginsenoside Ro	3.4
Glycyrrhizm	5.0
Thladioside H1 (1)	51

Table 4. Solubilizing effects on Yellow OB in buffer at 37°

Compound (1 mM)	Solubility of Yellow OB		
None	Insoluble		
Ginsenoside Ro	25.9 μM		
Thladioside H1 (1)	39 2 μM		

M/80 phosphate buffer (pH 65 ionic strength 0.02).

1714 R Nie et al

Compound (1 mg/ml)	Amount of saponin A (mg/ml)	Solubility of saponin A (mg/ml)	M/M*
None	1.6	0017	
Ginsenoside Ro	30	11	13
Thladioside H1 (1)	30	2 14	3 28

Table 5 Solubilizing effects on saponin A in water at 37°

was further purified by CC on silica gel (CHCl₃-MeOH-H₂O, 6 4:1) to give 1 in a yield of 9.7%.

Compound 1 a white powder, $[\alpha]_{D}^{22} + 3.8^{\circ}$ (H₂O, c 0.89), IR $v_{\rm max}$ cm⁻¹ 3300 (OH), 1740, 1720 (Found C, 54.32; H, 7.22. C₆₃H₃₈O₃₁ 2H₂O requires C, 54.53; H, 7.41%), ¹H NMR (270 MHz) δ 1 79 (3H, d, J = 5.9 Hz, H-6 of Rha), 4.92 (1H, d, J = 7.3 Hz, anomeric proton), 5.09 (1H, d, J = 7.0 Hz, anomeric proton), 5.24 (1H, d, J = 7.3 Hz, anomeric proton), 5.27 (1H, d, J = 8.4 Hz, anomeric proton), 5.42 (1H, t-like, H-12), 6.19 (1H, d, J = 6.6 Hz, anomeric proton of Xyl ester), 6.37 (1H, s, anomeric proton of Rha), 10.0 (1H, s, -CHO).

Enzymatic hydrolysis of compound 1 with crude hesperidinase. A soln of compound 1 (1 g) and crude hesperidinase (1 g, Tanabe Pharm. Co, Ltd, Osaka, Japan) in H_2O (120 ml) was incubated for 8 days at 37° After cooling, the reaction mixture was coned to dryness. The residue was chromatographed on silica gel (CHCl₃-MeOH, 49:1) to give the aglycone 2 (= gypsogenin) as colourless needles from aq. MeOH, mp 265–267° (uncorr.), [α] $_{20}^{20}$ (EtOH, c 0.99) which was identified by comparison of the melting point, optical rotation and ^{13}C NMR spectrum with those of the reported data [2]

Preparation of compound 3 from compound 1. NaBH₄ (1 1 g) was added to a soln of compound 1 (2.0 g) in H₂O (80 ml) and the mixture was stirred for 18 hr at room temp. Me₂CO (5 ml) was added to the reaction mixture and the solvent was evapn to dryness. The residue was passed through a highly porous polymer, DIAION HP-20 (Mitsubishi Chem Ind Tokyo, Japan) column (solvent. H₂O and MeOH successively). The MeOH eluate was evapd to dryness to give compound 3 (1 8 g) as a white powder, $[\alpha]_D^{12} - 13 \, 2^{\circ} \, (\text{H}_2\text{O}, c \, 0.97)$, IR $v_{\text{max}} \, \text{cm}^{-1} \, 3300 \, (-\text{OH})$, 1750 (ester), the ¹H NMR spectrum showed no -CHO signal (Found C, 55 00, H, 7 42 requires C₆₃H₁₀₀O₃₁·H₂O C, 55 17 H, 7 50%)

Selective cleavage of the ester glycoside linkage of 3. A soln of 3 (250 mg) and LiI (dried at 150° for 3 hr, 650 mg) and dry 2,6lutidine (10 ml) in dry MeOH (10 ml) was refluxed for 83 hr under Ar atmosphere. After cooling, the reaction mixture was diluted with 50% aq MeOH (3 ml), deionized with Amberlite MB-3 (H+, OH- form) and evapd to dryness. The residue was dissolved in H₂O and extd with n-BuOH. The H₂O layer was chromatographed on silica gel (CHCl₃-MeOH-H₂O, 10:5.1) to give 6 (64 mg) as a white powder which was subjected to the methylation analysis (2-linked, 3-linked and terminal xylopyranose residue and a 4-linked rhamnopyranose residue) The n-BuOH extract was chromatographed on silica gel (CHCl₃-MeOH-H₂O, 10.5 1 and then 6.4.1) to give compound 5 (81 mg) A soln of 5 (28 mg) in 2% HCl-MeOH (10 ml) was stirred for 2 hr at room temp. After neutralization with Amberlite MB-3, the solvent was evapd to dryness to give compound 7 as a white powder, $[\alpha]_D^{21} + 19.6^{\circ}$ (c = 0.75), IR v_{max} cm⁻¹ 3300 (OH), 1725 (ester), 1680 (COOH), ¹H NMR (100 MHz) δ 3 67 (3H, s, -COOMe of GlcUA), 5.12 (1H, d, J = 5 Hz, anomeric proton), 5 26 (1H, d, J = 8 Hz, anomeric proton). (Found: C, 58 90, H, 8 23. C₄₃H₆₈O₁₅ 3H₂O requires: C, 58.75, H, 8.49%)

Preparation of compound 8 and 8- 2 H₂ from compound 7 A soln of 7 (42 mg) and NaBH₄ (39 mg) in EtOH (3 ml) was stirred for 20 min at room temp. After decomposition of the excess of the reagent with Me₂CO (2 ml), the reaction mixture was deionized with Amberlite MB-3 and coned in vacuo to give compound 8 (29 mg) as a white powder, $[\alpha]_D^{21} + 32.0^{\circ}$ (c 1.10), IR v_{max} cm⁻¹ 3250 (OH), 1680 (COOH), 1 H NMR (100 MHz) δ 5.00 (1H, d, J = 6 Hz, anomeric proton), 5 24 (1H, d, J = 8 Hz, anomeric proton) (Found: C, 60 84, H, 8 66 C₄₂H₆₈O₁₄ 2H₂O requires C, 60 55, H, 8 71%). The deuterated compound, 8- 2 H₂ was prepared with NaB²H₄ in the same way The methylation analysis of 8- 2 H₂ indicated the presence of a 2-linked 6-dideuteroglucopyranosyl and a terminal galactopyranosyl residue.

Enzymatic hydrolysis with β -glucuronidase of compound 1 A soln of compound 1 (500 mg) and β -glucuronidase (3 ml, Sigma Co, from Helix pomatia) in H₂O (30 ml) was incubated for 44 hr at 37°. After cooling, the reaction mixture was chromatographed on DIAION HP-20 (H₂O and MeOH, successively) The MeOH cluate (294 mg) was chromatographed on silica gel (CHCl₃-MeOH-H₂O, 40·10 1 and then 30 10 1) to give compounds 9 (73 mg), 10 (40 mg) and 11 (41 mg) together with a mixture of 10 and 11 (70 mg).

Compound 9 A white powder, $[\alpha]_D^{22} + 10.3^\circ$ (c 0.80) IR v_{max} cm⁻¹ 3300 (OH), 1750, 1720 (ester), ¹H NMR (100 MHz) δ 1 75 (3H, d, J = 6 Hz, H-6 of Rha), 6 12 (1H, d, J = 8 Hz, anomeric proton of Xyl ester), 6 50 (1H, s, anomeric proton of Rha), 9 60 (1H, s, -CHO), (Found C, 62 82; H, 8 66. $C_{41}H_{64}O_{12} \cdot 2H_2O$ requires C, 62 73; H, 8 73%)

Compound 10 A white powder, $[\alpha]_D^{2^2} + 10.9^\circ$ (c 0.65). IR $v_{\rm max}{\rm cm}^{-1}$ 3300 (OH), 1750, 1720 (ester), ¹H NMR (100 MHz). δ 1.83 (3H, d, J=5 Hz, H-6 of Rha), 5.06 (1H, virtual coupling, anomeric proton of terminal Xyl), δ 16 (1H, d, J=6 Hz, anomeric proton of Xyl ester), δ 38 (1H, s, anomeric proton of Rha), 9.60 (1H, s, ~CHO). (Found C, 60.44, H, 8.33 C₄₆H₇₂O₁₆ 2H₂O requires C, 60.24, H, 8.35%)

Compound 11 A white powder, $[\alpha]_D^{22} - 4.89^{\circ}$ (c 0.47). IR v_{max} cm⁻¹· 3300 (OH), 1750, 1720 (ester), ¹H NMR (100 MHz) δ 1.83 (3H, d, J=5 Hz, H-6 of Rha), 506 (1H, virtual coupling, anomeric proton of terminal Xyl), 521 (1H, d, J=6 Hz, anomeric proton of inner Xyl), 616 (1H, d, J=6 Hz, anomeric proton of Xyl ester), 6.35 (1H, s, H-1 of Rha), 9.70 (1H, s, -CHO). (Found C, 58.13, H, 818 $C_{51}H_{80}O_{20}$ 2H₂O requires C, 58.38, H, 8.07%).

Selective cleavage of the ester glycoside linkage of compound 10. A soln of compound 10 (75 mg), dry LiI (102 mg) and dry 2,6-lutidine (3 ml) in dry MeOH (3 ml) was refluxed for 18 hr under Ar atmosphere. After cooling, the reaction mixture was diluted with 50% aq MeOH (2 ml), deionized with Amberlite MB-3 (H⁺, OH⁻ form) and evapn to dryness The residue was chromatographed on silica gel (CHCl₃-MeOH-H₂O 30 10 1) to give 11 (16 mg) as a white powder, which was subjected to

^{*} Molar ratio of solubilized saponin A to bisdesmosides

methylation analysis (2-linked and terminal xylopyranose units and 4-linked rhamnopyranose unit).

Solubilizing effect of compound 1 on saikosaponin-a, Yellow OB and saponin A. The effects were determined by the same procedure reported previously [11, 12]

Acknowledgements—We are grateful to Dr H. Matsuura (Wakunaga Pharm Co. Ltd, Hiroshima) for measurement of 270 MHz NMR. We also thank Dr H. Kanamori (Hiroshima Prefectural Institute of Public Health) for measurement of FD-MS. This study was financially supported by Grant-in-Aid for Overseas Scientific Research from the Ministry of Education, Science and Culture in 1988 (No. 63044100) to which the authors' thanks are

REFERENCE

- 1. Kohda, H. and Tanaka, O. (1975) Yakugaku Zasshi 95, 246.
- Tori, K., Seo, S., Shimada, A. and Tomita, Y. (1974) Tetrahedron Letters 48, 4227.

- 3. Kon, G. R., and Soper, H. R. (1940) J. Chem. Soc 617.
- 4. Kasai, R., Okihara, M., Asakawa, J., Mizutani, K. and Tanaka, O. (1979) Tetrahedron 35, 1427.
- Kizu, H. and Tomimori, T. (1982) Chem. Pharm. Bull 30, 3340.
- Ohtani, K., Mizutani, K., Kasai, R and Tanaka, O. (1984) Tetrahedron Letters 25, 4537.
- Kıtagawa, I., Yoshikawa, M., Yosioka, I. (1976) Chem. Pharm. Bull 24, 121.
- 8. Bjorndal, H, Lindberg, B., Pilotti, A. and Svensson, S. (1970) Carbohydr. Res. 15, 339.
- Watanabe, K., Fujino, H., Morita, T., Kasai, R. and Tanaka,
 O. (1988) Planta Med. 5, 377.
- 10 Sasakı, Y., Mızutanı, K., Kasai, R and Tanaka, O (1988) Chem. Pharm. Bull 36, 3491.
- Kimata, H., Sumida, N., Matsufuji, N., Morita, T., Ito, K., Yata, N. and Tanaka, O. (1985) Chem. Pharm. Bull. 33, 2849.
- Nakayama, K., Fujino, H., Kasai, R., Mitoma, Y., Yata, N. and Tanaka, O. (1986) Chem Pharm. Bull. 34, 3279.